

# Nanostructured tungsten trioxide thin films by aqueous chemical growth for applications in gas sensing and electrochromism

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## Abstract

Aqueous Chemical Growth (ACG) [1-3] is a low cost, low temperature and environmentally benign wet-chemistry technique that has been used to synthesize thin films and coatings of multi-functional Semiconductor Metal Oxides (SMO) that often find applications in gas sensing, smart windows, batteries, supercapacitors, etc.

We report here the use of the ACG technique to produce on bare Corning glass and F-doped Tin Oxide-on-glass (FTO) thin films of WO<sub>3</sub>, a SMO, which finds applications in gas sensing and electrochromic devices. SEM showed that nanoplatelet-containing structures were generally produced on the Corning glass substrates while urchin-like microspheres were produced on the FTO substrates. TEM, HRTEM, were used confirm the morphology of the structures observed in SEM while XRD alongside Raman spectroscopy were used to show that WO<sub>3</sub> in the thin films existed in the monoclinic, cubic and hexagonal phases.

While the WO<sub>3</sub> thin films prepared on Corning glass substrates were evaluated for their gas sensing behaviour with respect to hydrogen, CO, and CO<sub>2</sub> (flammable and poisonous gases common in mining and industrial environments), those that were prepared on FTO were evaluated for their electrochromic behaviour using Cyclic Voltammetry and UV-Vis-NIR spectrophotometry.

Results obtained on gas sensing showed that WO<sub>3</sub> thin films on Corning glass are suitable for hydrogen sensing in the 200-350 C temperature window (Fig.1). Doping these thin films with graphene resulted in reduction of sensing temperatures to 100 C. Gas sensing of CO and CO<sub>2</sub> was also observed to take place for the undoped WO<sub>3</sub> thin films at temperatures of 200 C and above.

For electrochromism (Fig. 2), the WO<sub>3</sub> thin films on FTO demonstrated fairly fast optical switching rates from blue to colourless, of less than 30 seconds upon H<sup>+</sup> intercalation in 0.1 M H<sub>2</sub>SO<sub>4</sub> electrolytic medium. This makes them potentially applicable for use as electrochromic materials in electronic displays, smart windows and other devices where optical switching is needed.

## References

- [1] Lionel Vayssieres, Anders Hagfeldt, and Sten Eric Lindquist, Pure Applied Chemistry, **72** (2000) 47-52.
- [2] Lionel Vayssieres and Arumugam Manthiram, J. Phys. Chem. B, **107** (2003) 2623-2625.
- [3] Lionel Vayssieres, Lew Rabenberg, and A. Manthiram, Nano Lett. , **2** (2002) 1393-1395.

## Figures

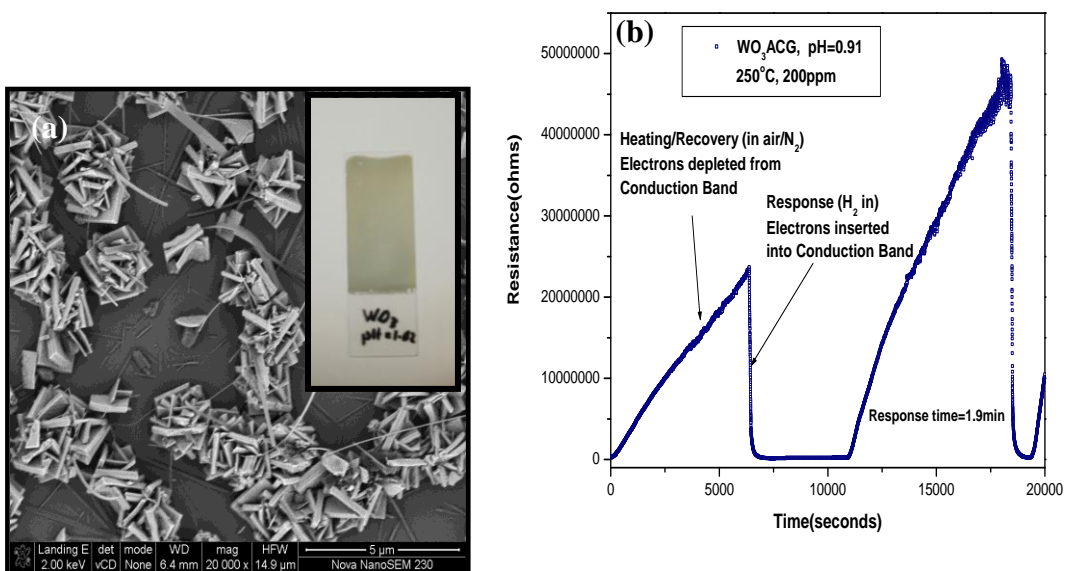


Fig.1. (a) SEM micrograph of WO<sub>3</sub> nanoplatelet-containing desert-rose like structures; (b) Hydrogen sensing of WO<sub>3</sub> thin film by Aqueous Chemical Growth at 250 C. Inset in Fig.1a shows WO<sub>3</sub> thin film on Corning glass.

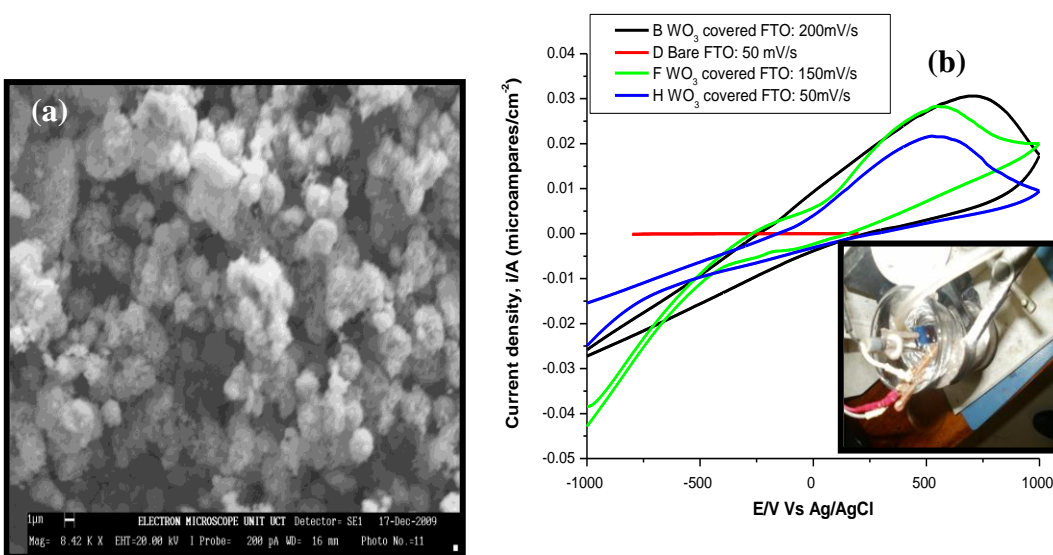


Fig.2. (a) SEM micrograph of WO<sub>3</sub> urchin-like microspheres produced by ACG; (b) Cyclic voltammogram of H<sup>+</sup> intercalation in WO<sub>3</sub> thin films on FTO carried out in 0.1M H<sub>2</sub>SO<sub>4</sub> medium. Inset in Fig. 2b shows electrochromic effect in WO<sub>3</sub> thin film on FTO.